

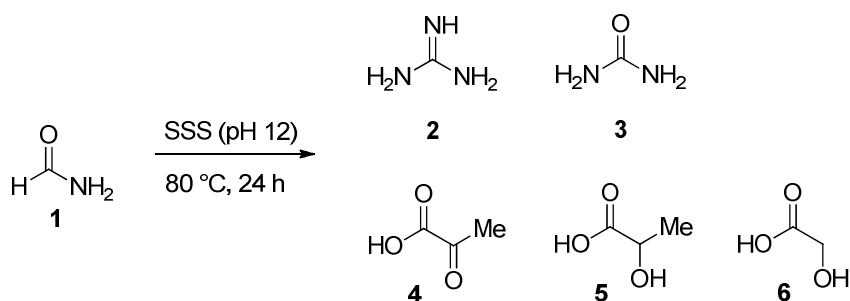
## SUPPORTING INFORMATION

**SI # 1 - Video 1.** Silica gardens of iron silicate hydrate formed in the presence of formamide at 2% (v/v) using a pellet of FeCl<sub>3</sub>.

### SI # 2 - Compounds obtained in the three experimental conditions

#### a) Control reaction: NH<sub>2</sub>CHO and SSS (pH 12).

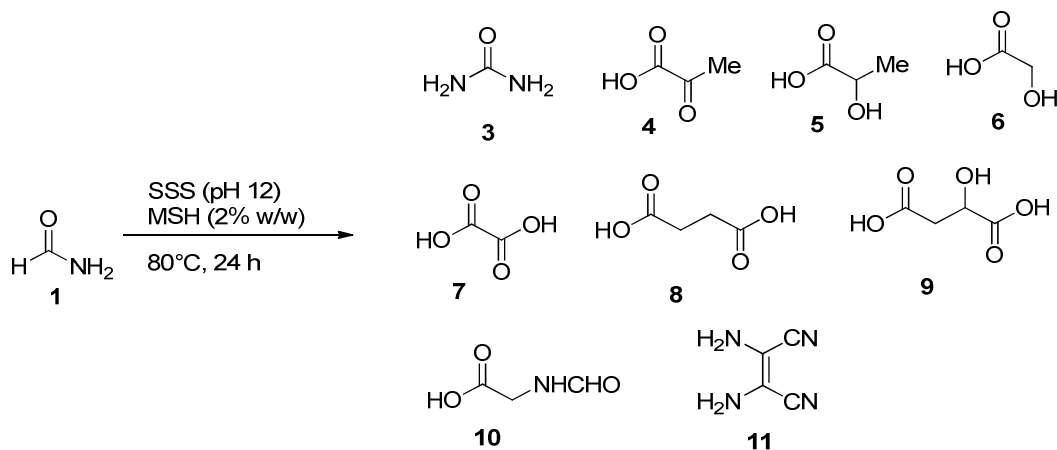
The reaction of NH<sub>2</sub>CHO with SSS (pH 12) was performed at three temperatures: 25°C, 80°C and 120°C for 24 h. Products are reported in Scheme 1. Few products were detected (in small or traces amounts) including guanidine **2**, urea **3**, pyruvic acid **4**, lactic acid **5**, and glycolic acid **6**. In particular, NH<sub>2</sub>CHO (200 µL) and SSS (2.0 mL; commercially available Aldrich solution) were heated at the appropriate temperature. At the end of the reaction the solution was distilled under high vacuum to remove the excess of NH<sub>2</sub>CHO and water. The residue was suspended in MeOH (2.0 mL) under magnetic stirring at 30°C for 30 minutes to separate the organic products from insoluble salt, and the organic layer was recovered by centrifugation. After evaporation of the MeOH layer, the residue was analysed by gas-chromatography associated to mass-spectrometry (GC-MS) after formation of the corresponding trimethylsilyl ethers (TMS). The analysis was limited to products present in yields ≥1 ng/mL. The yield was calculated as µg of product per mL of starting NH<sub>2</sub>CHO using oleic acid as internal standard. The data are the mean values of three experiments with standard deviation less than 0.1 %.



Scheme S1. Reaction of NH<sub>2</sub>CHO with sodium silicate solution (SSS) at 80 °C.

**b) Outer reaction:  $\text{NH}_2\text{CHO}$ , SSS (pH 12), and MSH.**

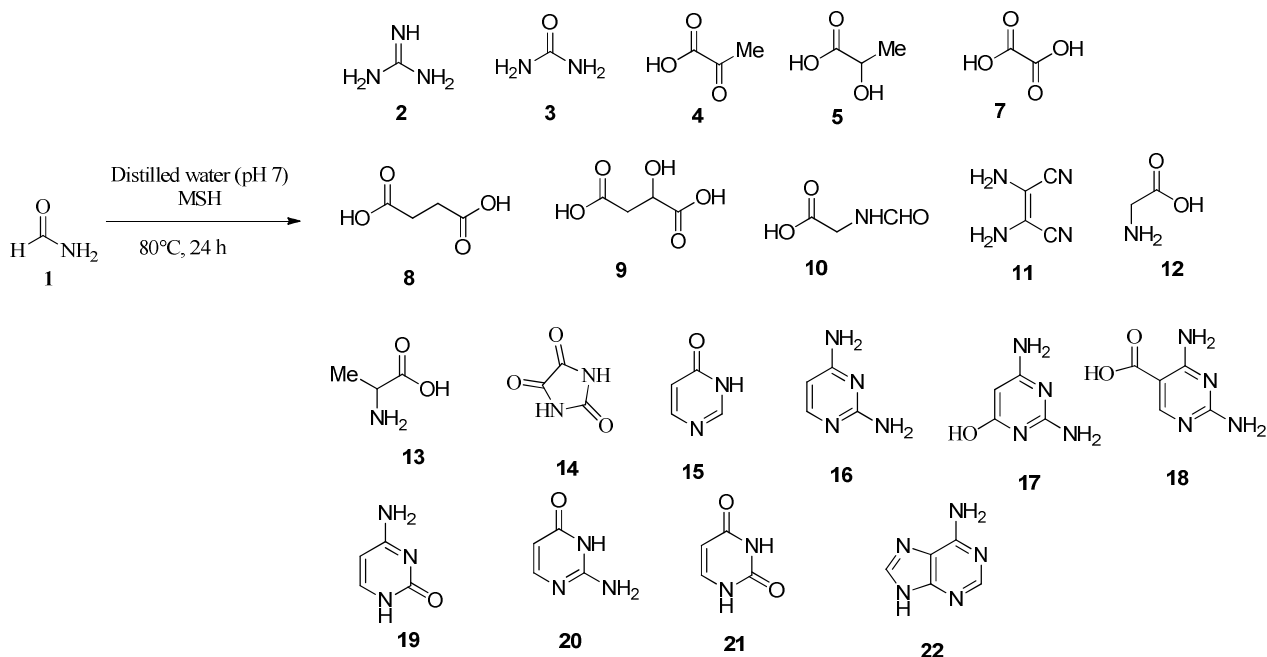
The reaction of  $\text{NH}_2\text{CHO}$  with SSS was performed in the presence of selected MSH at  $80^\circ\text{C}$  for 24 h. Products are reported in Scheme 2. In particular,  $\text{NH}_2\text{CHO}$  (200  $\mu\text{L}$ ) and SSS (2.0 mL; commercially available Aldrich solution) were heated at  $80^\circ\text{C}$  for 24 h in the presence of catalytic amount (2.0 % w/w) of the appropriate MSH [ $\text{ZnCl}_2$ ,  $\text{FeCl}_2 \cdot 4\text{H}_2\text{O}$ ,  $\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$ ,  $\text{Fe}_2(\text{SO}_4)_3 \cdot 9\text{H}_2\text{O}$ , and  $\text{MgSO}_4$ ]. At the end of the reaction the mixture was filtered to remove the inorganic membrane, followed by distillation under high vacuum to remove the excess of  $\text{NH}_2\text{CHO}$  and water. The residue was suspended in MeOH (2.0 mL) under magnetic stirring at  $30^\circ\text{C}$  for 30 minutes to separate the organic products from insoluble salt, and the organic layer was recovered by centrifugation. After evaporation of the MeOH layer, the residue was analysed by gas-chromatography associated to mass-spectrometry (GC-MS) after formation of the corresponding trimethylsilyl ethers (TMs). The analysis was limited to products present in yields  $\geq 1$  ng/mL. The yield was calculated as  $\mu\text{g}$  of product per mL of starting  $\text{NH}_2\text{CHO}$  using oleic acid as internal standard. The data are the mean values of three experiments with standard deviation less than 0.1 %. A panel of products larger than the reference reaction was detected, including different carboxylic acids, such as pyruvic **4**, lactic **5**, glycolic **6**, oxalic **7**, succinic **8**, and malic **9** acids, one amino acid *N*-formylglycine **10**, DAMN **11**, guanidine **2** and urea **3**.



Scheme S2. Reaction of  $\text{NH}_2\text{CHO}$  with sodium silicate solution (SSS) in the presence of MSH.

**c) Inner reaction: NH<sub>2</sub>CHO, distilled water (pH 7), and MSH.**

The reaction of NH<sub>2</sub>CHO with SSS was performed in distilled water (pH 7) in the presence of selected MSH at 80°C for 24 h. In particular, NH<sub>2</sub>CHO (200 µL) and distilled water (2.0 mL) were heated at 80°C for 24 h in the presence of catalytic amount (2.0 % w/w) of the appropriate MSH [ZnCl<sub>2</sub>, FeCl<sub>2</sub>•4H<sub>2</sub>O, CuCl<sub>2</sub>•2H<sub>2</sub>O, MnCl<sub>2</sub>•4H<sub>2</sub>O, Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>•9H<sub>2</sub>O, MgSO<sub>4</sub>, CuN<sub>2</sub>O<sub>6</sub>•3H<sub>2</sub>O]. At the end of the reaction the mixture was filtered to remove the inorganic membrane, followed by distillation under high vacuum to remove the excess of NH<sub>2</sub>CHO and water. The residue was analysed by gas-chromatography associated to mass-spectrometry (GC-MS) after formation of the corresponding trimethylsilyl ethers (TMs). The yield was calculated as µg of product per mL of starting NH<sub>2</sub>CHO using oleic acid as internal standard. The data are the mean values of three experiments with standard deviation less than 0.1 %. A large panel of product was obtained, including nucleobases cytosine **19**, uracil **21** and adenine **22**, nucleobase biosesters 2,4-diamino pyrimidine (2,4-DAP) **16** and isocytosine **20**, nucleobase analogues 4(3H)pyrimidinone **15**, 6-hydroxy-2,4-diamino pyrimidine **17** and 2,4-diamino pyrimidine-5-carboxylic acid **18**. Pyruvic acid **4**, lactic acid **5**, glycolic acid **6**, oxalic acid **7**, succinic acid **8**, malic acid **9**, *N*-formylglycine **10**, DAMN **11** and urea **3** were also detected.



Scheme S3. Reaction of NH<sub>2</sub>CHO in distilled water with MSH.

### SI # 3: Selected GC-MS chromatograms

#### GC-MS FeCl<sub>2</sub> outside

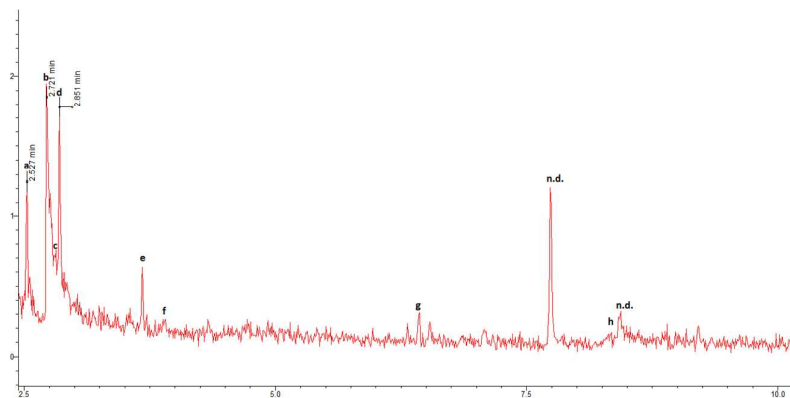


Figure S1: a, urea; b, guanidine; c, DAMN; d, glycolic acid; e, lactic acid; f, oxalic acid; g, pyruvic acid; h, formyl glycine; n.d., not determined.

#### GC-MS Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> outside

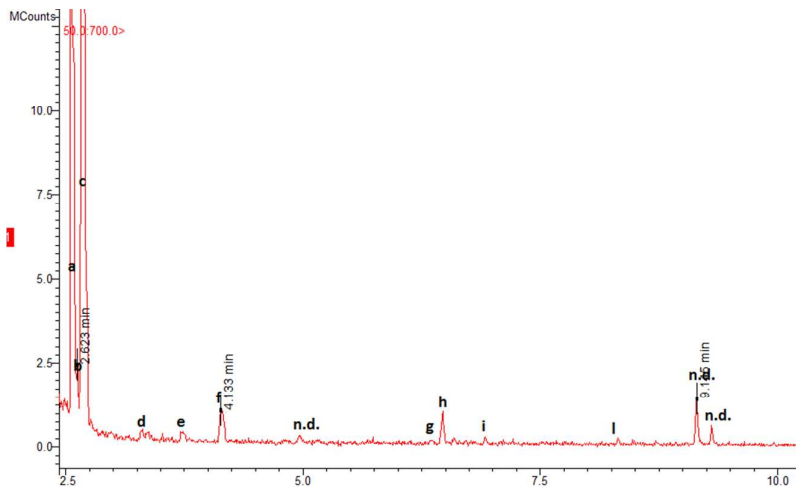


Figure S2: a, urea; b, DAMN; c, guanidine; d, glycolic acid; e, lactic acid; f, oxalic acid; g, pyruvic acid; h, succinic acid; i, malic acid; l, formyl glycine; n.d., not determined.

### GC-MS MgSO<sub>4</sub> inside

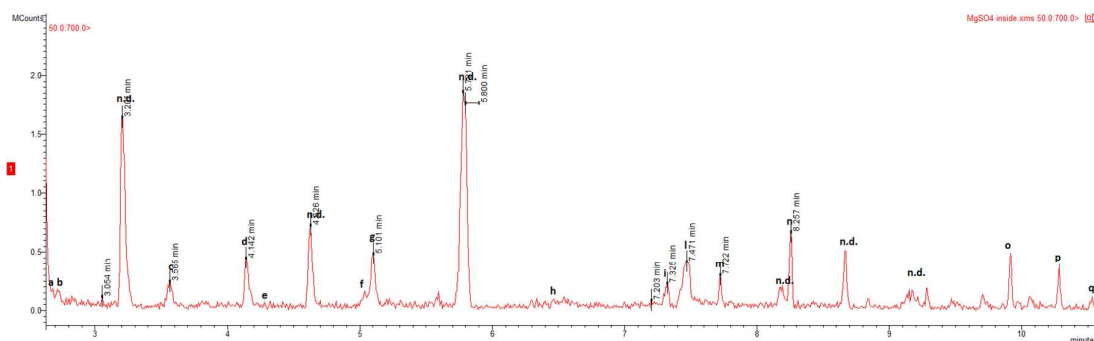


Figure S3: a, urea; b, guanidine; c, alanine; d, lactic acid; e, oxalic acid; f, 2,4-DAP; g, pyruvic acid; h, 6(OH)-2,4-DAP; i, succinic acid; j, glycine; k, uracil; l, formyl glycine; m, cytosine; n, isocytosine; o, adenine; p, adenine; q, adenine; n.d., not determined.

### GC-MS CuCl<sub>2</sub> inside

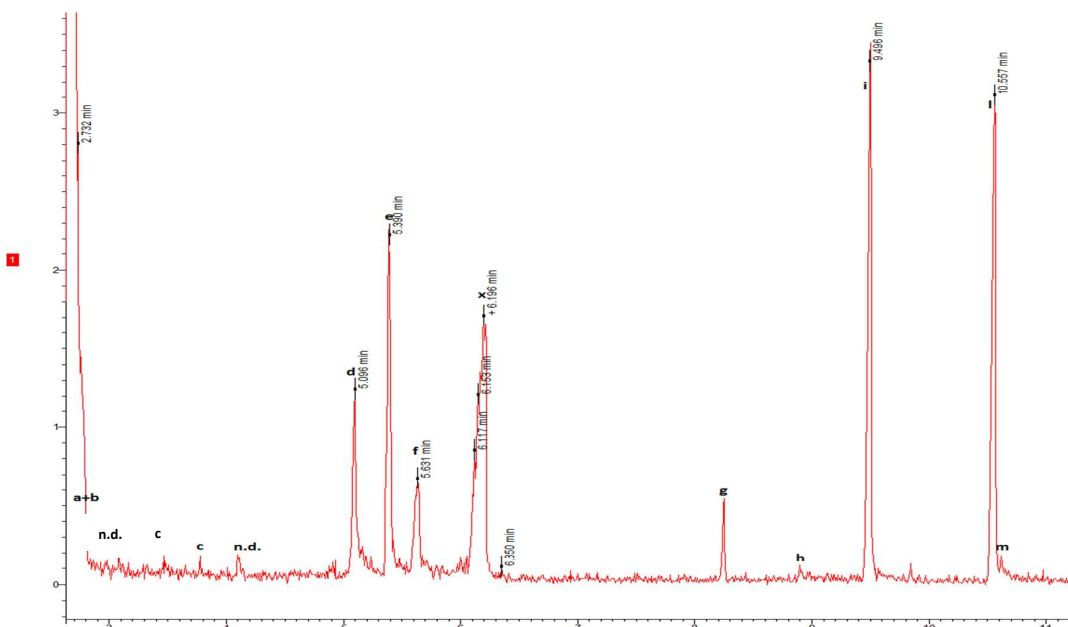


Figure S4: a, urea; b, guanidine; c, oxalic acid; d, pyruvic acid; e, 4(3H)-Pyr; f, glycine; g, silylating agent; h, parabanic acid; i, cytosine; j, uracil; k, isocytosine; l, adenine; m, adenine; n.d., not determined.

GC-MS CuN<sub>2</sub>O<sub>6</sub> inside

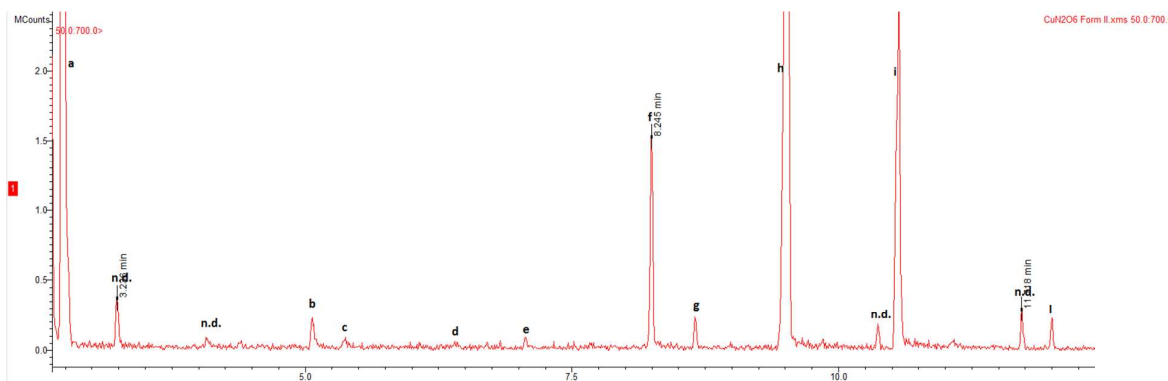


Figure S5: a, guanidine; b, 4(3H)-Pyr; c, pyruvic acid; d, 2,4-DAP; e, succinic acid; f, parabanic acid; g, cytosine; h, uracil; i, isocytosine; l, 2,4-DAP-5-COOH; m, n.d., not determined.

#### SI # 4: Mass-to-charge ratio (m/z) value and the abundance of mass spectra peaks of products

Table S1: Mass-to-charge ratio (m/z) value and the abundance of mass spectra peaks of compounds (2-22).

Products <sup>[a]</sup>	m/z (%)
Guanidine <sup>[c]</sup> (2)	188 (11) [M-CH <sub>3</sub> ], 173 (10) [M-2xCH <sub>3</sub> ], 171 (100), 73 (33)
Urea <sup>[c]</sup> (3)	204 (7) [M], 189 (73) [M-CH <sub>3</sub> ], 147 (100), 73 (35)
Piruvic acid <sup>[c]</sup> (4)	160 (10) [M], 145 (7) [M-CH <sub>3</sub> ], 88 (14) [M-Si(CH <sub>3</sub> ) <sub>3</sub> ], 71 (12) [M-Si(CH <sub>3</sub> ) <sub>3</sub> -OH], 43 (100) [M-HSi(CH <sub>3</sub> ) <sub>3</sub> -CO <sub>2</sub> ]
Lactic acid <sup>[c]</sup> (5)	219 (6) [M-CH <sub>3</sub> ], 190 (14) [M-CO <sub>2</sub> ], 147 (71) [M-Si(CH <sub>3</sub> ) <sub>3</sub> -CH <sub>3</sub> ], 133 (7), 117 (76) [M-Si(CH <sub>3</sub> ) <sub>3</sub> -(CH <sub>3</sub> ) <sub>3</sub> ]
Glycolic acid (6)	205 (8) [M-CH <sub>3</sub> ], 190 (1) [M-(CH <sub>3</sub> ) <sub>2</sub> ], 148 (10) [M-Si(CH <sub>3</sub> ) <sub>3</sub> ], 147 (74) [M-HSi(CH <sub>3</sub> ) <sub>3</sub> ], 133 (9) [M-CH <sub>3</sub> -Si(CH <sub>3</sub> ) <sub>3</sub> ], 117 (4) [M-(CH <sub>3</sub> ) <sub>2</sub> -HSi(CH <sub>3</sub> ) <sub>3</sub> ], 103 (5) [M-(CH <sub>3</sub> ) <sub>3</sub> -Si(CH <sub>3</sub> ) <sub>3</sub> ]
Oxalic acid <sup>[c]</sup> (7)	219 (3) [M-CH <sub>3</sub> ], 189 (5) [M-(CH <sub>3</sub> ) <sub>3</sub> ], 147 (78) [M-Si(CH <sub>3</sub> ) <sub>3</sub> -CH <sub>3</sub> ], 117 (1) [M-Si(CH <sub>3</sub> ) <sub>3</sub> -3xCH <sub>3</sub> ], 73 (100)
Succinic acid <sup>[c]</sup> (8)	247 (16) [M-CH <sub>3</sub> ], 173 (5) [M-HOSi(CH <sub>3</sub> ) <sub>3</sub> ], 147 (100), 73 (80)
Malic acid <sup>[d]</sup> (9)	350 (25) [M], 320 (9) [M-(CH <sub>3</sub> ) <sub>2</sub> ], 305 (15) [M-(CH <sub>3</sub> ) <sub>3</sub> ], 232 (70), 147 (98), 73 (100)
N-Formylglycine <sup>[b]</sup> (10)	160 (38) [M-CH <sub>3</sub> ], 147 (5) [M-CO], 131 (22) [M-CONH <sub>2</sub> ], 102 (11) [M-Si(CH <sub>3</sub> ) <sub>3</sub> ], 73 (100)
DAMN <sup>[c]</sup> (11)	252 (5) [M], 153 (18) [M-Si(CH <sub>3</sub> ) <sub>3</sub> -HCN], 138 (3) [M-NHSi(CH <sub>3</sub> ) <sub>3</sub> -HCN], 73 (100) [Si(CH <sub>3</sub> ) <sub>3</sub> ]
Glycine <sup>[b]</sup> (12)	147 (11) [M], 132 (28) [M-CH <sub>3</sub> ], 88 (9), 73 (100)
Alanine <sup>[c]</sup> (13)	218 (4) [M-CH <sub>3</sub> ], 190 (6), 147 (13), 116 (100) [M-HOSi(CH <sub>3</sub> ) <sub>3</sub> -CO], 73 (60)
Parabanic acid <sup>[c]</sup> (14)	258 (15) [M], 243 (35) [M-CH <sub>3</sub> ], 215 [M-2xCH <sub>3</sub> ], 100 (100), 73 (23)
4(3H)pyrimidinone <sup>[b]</sup> (15)	168 (25) [M], 153 (100) [M-CH <sub>3</sub> ], 123 (5) [M-(CH <sub>3</sub> ) <sub>3</sub> ], 99 (100)
2,4-DAP <sup>[c]</sup> (16)	366 (5) [M], 351 (100) [M-CH <sub>3</sub> ], 336 (12) [M-(CH <sub>3</sub> ) <sub>2</sub> ], 294 (5) [M-Si(CH <sub>3</sub> ) <sub>3</sub> ], 279 (8) [M-Si(CH <sub>3</sub> ) <sub>3</sub> -CH <sub>3</sub> ], 263 (15) [M-Si(CH <sub>3</sub> ) <sub>3</sub> -(CH <sub>3</sub> ) <sub>2</sub> -H], 221 (3) [M-Si(CH <sub>3</sub> ) <sub>3</sub> -HSi(CH <sub>3</sub> ) <sub>3</sub> ], 73 (70) [Si(CH <sub>3</sub> ) <sub>3</sub> ]
6(OH)-2,4-DAP <sup>[c]</sup> (17)	270 (35) [M], 255 (100) [M-CH <sub>3</sub> ]
2,4-DAP-5COOH <sup>[d]</sup> (18)	370 (11) [M], 355 (100) [M-CH <sub>3</sub> ]
Cytosine <sup>[c]</sup> (19)	255 (49) [M], 254 (100) [M-H], 240 (72) [M-CH <sub>3</sub> ], 182 (5) [M-HSi(CH <sub>3</sub> ) <sub>3</sub> ]
Isocytosine <sup>[d]</sup> (20)	327 (18) [M], 312 (100) [M-CH <sub>3</sub> ], 282 (9) [M-(CH <sub>3</sub> ) <sub>3</sub> ], 255 (6) [M-Si(CH <sub>3</sub> ) <sub>3</sub> ], 240 (7) [M-HSi(CH <sub>3</sub> ) <sub>3</sub> -CH <sub>3</sub> ], 183 (2) [M-2xSi(CH <sub>3</sub> ) <sub>3</sub> ]
Uracil <sup>[c]</sup> (21)	256 (35) [M], 241 (100) [M-CH <sub>3</sub> ], 225 (15) [M-CH <sub>3</sub> -CH <sub>4</sub> ], 182 (7) [M-Si(CH <sub>3</sub> ) <sub>3</sub> -H <sub>2</sub> ], 142 (70), 113 (55)
Adenine <sup>[c]</sup> (22)	279 (27) [M], 264 (100) [M-CH <sub>3</sub> ], 249 (1) [M-(CH <sub>3</sub> ) <sub>2</sub> ], 192 (17)

[a] Mass spectroscopy was performed by using a GC-MS. Samples were analyzed after treatment with *N,N*-bis-trimethylsilyltrifluoroacetamide and pyridine. The peak abundance is reported in parenthesis. [b] Product analyzed as the monosilyl derivative; [c] Product analyzed as the bis-silyl derivative; [d] Product analyzed as the tris-silyl derivative.

**SI # 5: Selected spectra of compounds identified in MSH catalysed synthesis according to Figure 2. All products have been recognized with a similarity index (S.I.) greater than 98% compared to reference standards**

Figure S6: Selected spectra of compounds identified in MSH catalysed synthesis.

